MATERIAL SAFETY DATA SHEET

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### Leisure Time OZ

1. Product And Company Identification				
Supplier Leisure Time 1400 Bluegrass Lakes Parkway Alpharetta, GA 30004 United States Telephone Number: (770) 521-5999 FAX Number: (770) 521-5959 Web Site: www.poolspacare.com	1400 Bluegrass Lakes F Alpharetta, GA 30004 Telephone Number: (7 FAX Number: (770) 521	Manufacturer Advantis Technologies, Inc. 1400 Bluegrass Lakes Parkway Alpharetta, GA 30004 United States Telephone Number: (770) 521-5999 FAX Number: (770) 521-5959 Web Site: www.poolspacare.com		
Supplier Emergency Contacts & Phone Number CHEMTREC - DAY OR NIGHT: (800) 424-9300	Manufacturer Emergency Contacts & Phone Number CHEMTREC - DAY OR NIGHT: (800) 424-9300			
Issue Date: 08/05/2004 Product Name: Leisure Time OZ Chemical Name: Potassium Monopersulfate Chemical Family: Peroxygen Salt Chemical Formula: Proprietary MSDS Number: 318				
2. Composition/Information On Ingredients				
Ingredient Name		CAS Number		Percent Of Total Weight
BORICACID		10043-35-3		
MAGNESIUMCARBONATE		546-93-0		
POTASSIUMBISULFATE		7646-93-7		
POTASSIUMPEROXYDISULFATE		7727-21-1		
POTASSIUMPEROXYMONOSULFATE		10058-23-8		
POTASSIUMSULFATE				
Sodium Carbonate		497-19-8		
Ingredients listed in this section have been determined to determined to be health hazards are listed if they compris carcinogens are listed if they comprise 0.1% or more of th 29CFR 1910.1200(i)(1).	e 1% or more of the composit	on. Materia	ls id	entified as
This product is a skin and eye corrosive, and a nose a				
Hazards Identification (Pictograms)				
3. Hazards Identification				
Primary Routes(s) Of Entry				
Skin Contact				

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#### 3. Hazards Identification - Continued

#### Eye Hazards

Corrosive to eyes. Eye contact may cause corrosion or ulceration. Severe eye damage may result if not immediately treated.

#### Skin Hazards

Corrosive to skin. Skin contact with aqueous solutions or the dry powder upon contact with moisture or perspiration may cause skin burns or ulceration; temporary body hair loss may occur in contacted areas. Skin contact with the product may cause allergic skin reactions in sensitive individuals.

#### **Ingestion Hazards**

May cause irritation of the throat. Ingestion may cause inflammation and damage to the lining of the stomach, resulting in bleeding. Ingestion may cause gastritis possibly progressing to necrosis or hemorrhage.

#### Inhalation Hazards

Inhalation may cause nose bleeds and irritation of the upper respiratory passages with coughing and discomfort.

#### Signs And Symptoms

Irritant to eyes and skin due to oxidizing properties

#### First Aid (Pictograms)



#### 4. First Aid Measures

#### Eye

In case of contact, hold eyelids apart and immediately flush eyes with plenty of water for at least 15 minutes. Call a physician or a poison control center immediately.

#### <u>Skin</u>

Wash affected areas with soap and water. Call a physician. Wash clothing before reuse.

#### **Ingestion**

DO NOT INDUCE VOMITING. Drink large amounts of water. Call a physician or a poison control center immediately.

#### Inhalation

If inhaled, remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Consult Physician

#### Fire Fighting (Pictograms)



#### 5. Fire Fighting Measures

Flash Point: N/A °F

#### Fire And Explosion Hazards

Improper storage of large masses of this product can trap heat and lead to ignition of combustibles. Grinding or intensive mixing may cause decomposition with liberation of heat and oxygen; ignition of oxidizable material if present may occur.

#### Extinguishing Media

In case of fire, soak (flood) with water.

#### Fire Fighting Instructions

Will release oxygen when heated, intensifying a fire. Firefighters should wear self-contained breathing apparatus and full protective gear.

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5. Fire Fighting Measures - Continued

Fire Fighting Instructions - Continued

GRINDING OR INTENSIVE MIXING MAY GENERATE SUFFICIENT HEAT TO FUSE PRODUCT AND CAUSE IGNITION OF OXIDIZABLE MATERIAL PRESENT.

#### 6. Accidental Release Measures

Clean up spill immediately. Flush spill area with water in compliance with State and Federal Regulations.

Handling & Storage (Pictograms)



#### 7. Handling And Storage

#### Handling And Storage Precautions

Keep out of reach of children. Store material in a cool and dry place.

#### **Handling Precautions**

Avoid breathing dust or vapor. Avoid contact with skin and clothing. Avoid contact with eyes. Wash thoroughly after handling. Wash clothing after use.

#### Storage Precautions

Store in a cool dry place. Keep away from heat, sparks, flame, and other sources of ignition (i.e., pilot lights, electric motors and static electricity).

#### Work/Hygienic Practices

Use safe chemical handling procedures suitable for the hazards presended by this material.

#### Protective Clothing (Pictograms)



#### 8. Exposure Controls/Personal Protection

#### Engineering Controls

Local exhaust acceptable. Special exhaust not required

#### Eye/Face Protection

Safety glasses with side shields or goggles recommended.

#### Skin Protection

Chemical-resistant gloves.

#### **Respiratory Protection**

The level of respiratory protection needed should be based on the required protection factor after evaluating chemical exposures using appropriate industrial hygiene monitoring and/or OSHA guidance.

#### 9. Physical And Chemical Properties

#### <u>Appearance</u>

White tablet

<u>Odor</u> None

Chemical Type: Mixture Physical State: Solid Melting Point: DECOMPOSES °F

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#### 9. Physical And Chemical Properties - Continued

Odor - Continued Boiling Point: DECOMPOSES °F Specific Gravity: 1.2 Molecular Weight: PROPRIETARY Percent Volitales: NIL Vapor Pressure: NIL Vapor Density: NOT VOLATILE Solubility: >25% AT 20C (68F) Evaporation Rate: NOT VOLATILE Corrosive, Oxidizer

#### 10. Stability And Reactivity

#### Stability: Stable

Hazardous Polymerization: Will not occur

#### Conditions To Avoid (Stability)

The mixture of this product with compounds containing halides or active halogens (bromine, chlorine, iodine) can cause the release of the respective halogen gas, if moisture is present. Avoid these gases (bromine and chlorine) because they are very irritating to eyes and lungs even at low concentrations. Never mix concentrated product with dry or concentrated bromine containing chemicals, such as bromates, bromides, or any concentrated bromine pool chemicals. Mixing this product with dry or concentrated chlorine containing chemicals, such as hypochlorites, sodium dichloroisocyanurate, sodium triisocyanurate or with sodium chloride, may cause the release of chlorine gas.

Mixing with cyanides can cause release of hydrogen cyanide gas. Mixing with heavy metal salts such as those of cobalt, nickel, copper, or manganese can cause decomposition with release of oxygen and heat.

#### Incompatible Materials

Alkalis, Heavy Metals cause evolution of Oxygen gas. Halogens and cyanides can cause release of gases of these compounds.

#### Hazardous Decomposition Products

Oxygen

#### **11. Toxicological Information**

#### Acute Studies

This product is a severe skin and eye irritant, but is not a skin sensitizer in animals.

#### 12. Ecological Information

No Data Available...

#### 13. Disposal Considerations

Dispose in accordance with applicable federal, state and local government regulations.

#### 14. Transport Information

#### Proper Shipping Name

CORROSIVE SOLID, ACIDIC, INORGANIC, N.O.S. (Monopersulfate Compound)

#### Hazard Class

8, PGII (<=1kg Consumer Commodity ORM-D)

### DOT Identification Number

UN3260

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DOT (Pictograms)
15. Regulatory Information
SARA Hazard Classes Acute Health Hazard
Canadian Regulatory Information Class D, Div 2b - Toxic Material. Skin or Eye Irritant Class E - Corrosive Material Class C - Oxidizing Material
WHMIS - Canada (Pictograms)
NFPA HMIS
HEALTH 3 FLAMMABILITY 0 REACTIVITY 1 PERSONAL PROTECTION E
16. Other Information
Revision/Preparer Information MSDS Preparer: JHW This MSDS Superceeds A Previous MSDS Dated: 10/13/2003
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Leisure Time

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